



SAFETY DATA SHEET for OZONE

(Formerly MSDS)

1. PRODUCT IDENTIFICATION Product

Name: Ozone**Common Names/Synonyms:** Triatomic Oxygen, Trioxygen, O₃**Ozone Generator Manufacturer/Supplier**Pacific Water Technology Pty Ltd website: www.pacificwater.com.au

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Product Use: This SDS is limited to ozone produced in gaseous form on site by an ozone generator, in varying concentrations in either air or aqueous solution, for the purposes of odour abatement, oxidation of organic compounds, or antimicrobial intervention, in a variety of applications, from food processing to ground water remediation.

2. HAZARD IDENTIFICATION



GHS Classifications:

Physical Hazards	Health Hazards	Environmental Hazards
Oxidising Gas	Skin Irritation – Category 3	Acute Aquatic Toxicity – Category I
	Eye Irritation – Category 2B	
	Respiratory Systemic Toxicity – Category 1 (Acute & Repeated Exposures)	

NOTE: Severe respiratory toxicity will develop before skin or eye irritation go beyond listed categories. ***Anyone with chronic pulmonary problems, especially asthma, should avoid exposure to ozone.***

WHMIS Classifications (Workplace Hazardous Materials Information System, Canada)

D1A	Acute lethality – Very toxic, immediately	C	Oxidising
D2A	Chronic Toxicity – Very Toxic	F	Dangerously Reactive
D2B	Mutagenicity – Toxic		

Signal word (GHS-US)

Danger



Hazard statements (GHS-US)

H270 - May cause or intensify fire; oxidiser
H320 - Causes eye irritation
H370 - Causes damage to organs
H372 - Causes damage to organs through prolonged or repeated exposure
H400 - Very toxic to aquatic life

**WORK SAFE AUSTRALIA LABELLING
OF WORKPLACE
HAZARDOUS CHEMICALS**

P220 - Keep/Store away from clothing /.../combustible materials
P244 - Keep reduction valves/valves and fittings free from oil and grease
P260 - Do not breathe dust/fume/gas/mist/vapours/spray
P264 - Wash ... thoroughly after handling
P270 - Do not eat, drink or smoke when using this product
P273 - Avoid release to the environment
P305+P351+P338 - If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing
P307 + P311 - If exposed: Call a poison centre/doctor
P314 - Get medical advice/attention if you feel unwell
P321 - Specific treatment (see ... on this label)
Reference to supplemental first aid instruction.
- Manufacturer/supplier or the competent authority may specify a cleansing agent if appropriate.
P337+P313 - If eye irritation persists: Get medical advice/attention
P370+P376 - In case of fire: Stop leak if safe to do so
P391 - Collect spillage
P403 - Store in a well-ventilated place
P405 - Store locked up
P501 - Dispose of contents/container to ...in accordance with local/ regional/national/international Regulations (to be specified).

3. COMPOSITION

Chemical Formula O₃
CAS Registry Number 10028-15-6

**4. FIRST AID MEASURES**

Route of Entry		Symptoms	First Aid
Skin Contact	YES	Irritation	Rinse with water
Skin Absorption	NO	NA	NA
Eye Contact	YES	Irritation	Rinse with water, remove contacts
Ingestion	NO	NA	NA
Inhalation	YES	Headache, cough, dry throat, heavy chest, shortness of breath	Remove to fresh air, provide oxygen therapy as needed

For severe cases, or when symptoms don't improve, seek medical help.

5. FIRE FIGHTING MEASURES

While ozone itself is not flammable, it is a strong oxidant and may accelerate, even initiate, combustion, or cause explosions. Use whatever extinguishing agents are indicated for burning materials.

6. ACCIDENTAL RELEASE MEASURES

Turn off ozone generator, and ventilate the area. Evacuate the area until ozone levels subside.

7. HANDLING AND STORAGE

Ozone must be contained within ozone-resistant tubing and pipes from the generation point to the application point. Any leaks must be repaired before further use.

8. EXPOSURE CONTROLS/PERSONAL PROTECTION

The Environmental Protection (Air) Policy 2019 (EPP Air) objectives for ozone are:

- 0.10 parts per million (ppm) for a 1-hour exposure period
- 0.08ppm for a 4-hour exposure period.

The National Environment Protection (Ambient Air Quality) Measure standard for ozone is:

- 0.065ppm for an 8-hour exposure period

These guidelines are designed to keep ozone concentrations below the levels that affect sensitive members of the population.

NIOSH: Exposure Limit Ceiling Value **0.1 ppm** light; **0.08 ppm** moderate; **0.05 ppm**, heavy;

Light, moderate, heavy work TWA \leq 2 hours, **.2 ppm**

Immediately Dangerous to Life or Health **5 ppm**

Respiratory Protection: Use full face self-contained breathing apparatus for entering areas with high concentration of ozone.

Engineering controls: use ozone destruct units (thermal and/or catalytic) for off gassing ozone.

**9. PHYSICAL AND CHEMICAL PROPERTIES**

Physical state	Gas	pH	NA
Molecular Weight	48.0	Flash point	NA
Appearance	Clear at low concentrations, blue at higher concentrations	Evaporation rate	NA
Odour	Distinctive pungent odour	Flammability	NA
Odour threshold	0.02 to 0.05 ppm; exposure desensitises	Explosive limits	NA
Melting point	-193°C/-315°F	Relative density	NA
Boiling point	-112°C/-169°F	Partition coefficient	NA
Vapour pressure	> 1 atm	Auto-ignition temperature	NA
Vapour density	1.6 (air = 1)	Decomposition temperature	NA
Solubility in water	570 mg/L @20°C, 100% O ₃ ; 0.64 @0°C (vol/vol)	Viscosity	NA

10. STABILITY AND REACTIVITY

Ozone is highly unstable and highly reactive. Avoid contact with oxidisable substances, including alkenes, benzene and other aromatic compounds, rubber, Di- cyanogen, bromine diethyl ether, dinitrogen tetroxide, nitrogen trichloride, hydrogen bromide, and tetrafluorohydrazine. Ozone will readily react and spontaneously decompose under normal ambient temperatures.

11. TOXICOLOGICAL INFORMATION

Likely routes of exposure: inhalation, eyes, skin exposure.

Effects of Acute Exposure: Discomfort, including headache, coughing, dry throat, shortness of breath, heavy feeling in chest (including possible pulmonary oedema/fluid in the lungs); higher levels of exposure intensify symptoms. Irritation of skin and/or eyes is also possible.

Effects of Chronic Exposure: Similar to acute exposure effects, with possible development of chronic breathing disorders, including asthma.

Inhalation LC₅₀: mice, 12.6 ppm for 3 hours; hamsters, 35.5 ppm for 3 hours

Irritancy of Ozone	YES
Sensitisation to Ozone	NO
Carcinogenicity (NTP, IARC, OSHA)	NO
Reproductive Toxicity	Not Proven
Teratogenicity	Not Proven
Mutagenicity	Not Proven
Toxicologically Synergistic Products	Increase susceptibility to allergens, pathogens, irritants



12. ECOLOGICAL INFORMATION

The immediate surrounding area may be adversely affected by an ozone release, particularly plant life. Discharge of ozone in water solution would also be harmful to any aquatic life. Due to natural decomposition, bioaccumulation will not occur, and the area affected would be limited.

13. DISPOSAL CONSIDERATIONS

Off-gassing of ozone should be through an ozone destruct unit which uses heat and/or a catalyst to accomplish the breakdown of ozone to oxygen before release into the atmosphere.

14. TRANSPORT INFORMATION

NOT APPLICABLE, as ozone is unstable and either reacts with other substances in the environment or decomposes, and therefore must be generated at the location and time of use.

15. REGULATORY INFORMATION

SARA = Superfund Amendments and Renewal Act

SARA Title III Section 302 Extremely Hazardous Substance TPQ: 100 lbs.

SARA Title III Section 304, EHS RQ: 100 lbs.

SARA Title III Section 313: Ozone is reportable if more than 10,000 lbs. are used/year. TPQ (Threshold Planning Quantity) requires emergency planning activities if this amount is on site at any time during year

RQ (Reportable Quantity) requires any release of this amount into the environment to be reported to the National Response Centre

Source: EPA List of Lists

16. OTHER INFORMATION

The half-life of ozone is much shorter in water than in air. Increased temperature in either solvent decreases the half-life. Published research indicates a half-life of 20 minutes for ozone dissolved in water at 20°C, and a half-life of approximately 25 hours for ozone in dry air at 24°C (McClurkin & Maier, 2010). The practical half-life time is actually less, especially in air, due to air circulation, humidity, the presence of contaminants or walls with which to react, etc. In many situations, with air movement, warmer temperatures, and normal relative humidity, the half-life of ozone in air could be 1 hour or less. Further, ventilation of a closed space to other areas will also disperse the ozone, so that concentration levels can rapidly decrease after generation ceases.

Source websites:

Canadian Centre for Occupational Health and Safety: Chemical Profiles: Ozone

http://www.ccohs.ca/oshanswers/chemicals/chem_profiles/ozone/

Haz-Map: Occupational Exposure to Hazardous Agents: Ozone

<http://hazmap.nlm.nih.gov/category-details?id=68&table=copytblagents>

International Chemical Safety Cards #0068: Ozone

<http://www.cdc.gov/niosh/ipcsneng/neng0068.html>



NIOSH Pocket Guide to Chemical Hazards: Ozone <http://www.cdc.gov/niosh/npg/npgd0476.html>

United States National Library of Medicine ChemIDplus Lite: Ozone 10028-15-6

<http://chem.sis.nlm.nih.gov/chemidplus/ProxyServlet?objectHandle=DBMaint&actionHandle=default&nextPage=jsp/chemidlite/ResultScreen.jsp&TXTSUPERLISTID=0010028156>

Safe Work Australia. Code of Practice: Preparation of Safety Data Sheets for Hazardous Chemicals. 2016,

https://www.safeworkaustralia.gov.au/system/files/documents/1902/code_of_practice_preparation_of_safety_data_sheets_for_hazardous_chemicals_0.pdf.

Safe Work Australia. (2013). Guidance on the interpretation of workplace exposure standards for airborne contaminants (2nd ed.). Retrieved from

<https://www.safeworkaustralia.gov.au/system/files/documents/1705/guidance-interpretation-workplace-exposure-standards-airborne-contaminants-v2.pdf>

Queensland Government. (n.d.). *Ozone*. Retrieved from

<https://www.qld.gov.au/environment/management/monitoring/air/air-pollution/pollutants/ozone>

Work Safe Australia Labelling Of Workplace Hazardous Chemicals:

<https://www.safeworkaustralia.gov.au/doc/model-code-practice-labelling-workplace-hazardous-chemicals>

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